Some Properties of Re₂Te₅-based Materials

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Abstract

Re2Tes is a semiconducting compound with an energy band gap of about 0.8 eV. It has a relatively complex crystal structure with 84 atoms per unit cell. Initial results obtained at the Jet Propulsion Laboratory on p-type polycrystalline samples showed that they possess large Seebeck coefficient values but also large electrical resistivity values. They also exhibit very low thermal conductivity with a room temperature value of 13 mW/cmK. Another attractive feature of Re₂Te₅ is the possibility of inserting a variety of atoms in the large voids (2.8 Å in diameter) of the crystal structure to form Re₆M₂Te₁₅ filled compositions. The void fillers could act as phonon scattering centers, further reducing the thermal conductivity in these materials. As part of an effort to evaluate the potential of Re₂Te₅-based materials for thermoelectric applications, we are currently exploring the synthesis and properties of filled compositions as well as n-type Re₂Te₅ samples. We present and discuss in this paper initial results obtained on Fe and Ag doped Re₂Te₅ samples.

Introduction

One of the new approaches for developing thermoelectric materials with superior figures of merit is to look at materials which can be referred to as "rattling" semiconductors. Originally proposed by Slack [1], the idea is that, in crystals containing loosely bound atoms, phonons should be scattered more strongly than electrons (holes). Such materials should, therefore possess both a low thermal conductivity and good electrical properties (high carrier mobility). Low lattice thermal conductivity have been measured for compounds from at least two classes of "rattling" materials: skutterudites and Chevrel compounds [2-6].

Another compound with a crystal structure that can host "rattling" atoms is Re₆Te₁₅ (Re₂Te₅). The crystal structure of the cluster compound Re₆Te₁₅ was studied in details by Klaiber et al. [7]. This compound, with 84 atoms per unit cell, belongs to the space group Pbca with a=13.003Å, b=12.935Å and c=14.212Å. The crystal structure presents some similarities with the Chevrel phases and the Re atoms are also arranged in octahedral [Re₆] clusters (see Figure 1). A number of voids are present in the structure and are illustrated on Figure 1. The large spheres represent the atoms that can possibly be inserted in these voids. The radius of the voids is 2.754Å [7] and therefore each of the voids is large enough to accommodate a great number of different type of atoms. Filled compositions can be represented by the formula Re₆M₂Te₁₅. Information about the thermoelectric properties of Re₆Te₁₅ is rather limited in the literature and we have started to synthesize and measure the properties of Re₆Te₁₅-based materials. Initial results, reported in an earlier publication [7], show that, as expected from the heavy masses of the elements forming the compound as well as the large number of atoms per unit cell, the thermal conductivity values for Re_6Te_{15} are low (~13 mW/cmK at 300K). Undoped samples are p-type and possess high Seebeck coefficient and electrical resistivity. The lattice thermal conductivity is reduced further by point defect scattering in $Re_6Te_{15-x}Se_x$ solid solutions [6]. In addition, it was found that several filled compositions $Re_6Ag_xTe_{15}$ with $0.5 \le x \le 1.14$ could be formed. In an effort to further characterize the properties of Re_6Te_{15} -based materials, we have synthesized Fe and Ag doped samples and measured some of their properties. The results are reported and discussed in this paper.

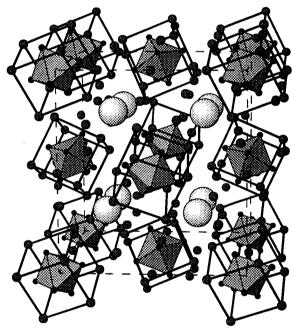


Figure 1: Illustration of the Re₆Te₁₅ unit cell showing the [Re₆] cluster surrounded by eight Te atoms. Large spheres represent atoms inserted in the voids. Some Te atoms were omitted for clarity.

Experimental

Polycrystalline samples of Ag- and Fe-doped Re₆Te₁₅ were prepared by mixing and reacting elemental powders of rhenium (99.997%), tellurium (99.999%) and iron (99.999%) or silver (99.99%). The nominal compositions were Re_{5.5}Fe_{0.5}Te₁₅ and Re₆Te_{14.5}Ag_{0.5}. The powders were first mechanically mixed in a plastic vial before being loaded into quartz ampoules which were evacuated and sealed. The ampoules were then heated at 973K for 10 days with one intermediate crushing. The samples were analyzed by x-ray diffractometry (XRD) to check that they were single phase.

The powders were then hot-pressed in graphite dies into dense samples, 10 mm long and 12 mm in diameter. The hot-pressing was conducted at a pressure of about 20,000 psi and at a temperature of 973 K for about 2 hours under argon atmosphere. The density of the samples was calculated from the measured weight and dimensions and was found to be about 97% of the theoretical density.

The samples were also characterized by microprobe analysis (MPA) which was performed using a JEOL JXA-733 electron superprobe operating at $20x10^3$ Volts (V) of accelerating potential and $15x10^{-9}$ Amperes (A) of probe current. All samples were found to contain less than 1% of secondary phase by microprobe analysis.

Samples in the form of disks (typically a 1.0 mm thick, 12 mm diameter slice) were cut from the cylinders using a diamond saw (perpendicular to the pressing direction) for electrical and thermal transport property measurements. All samples were characterized at room temperature by Seebeck coefficient, Hall effect and electrical resistivity measurements. Temperature dependence of electrical resistivity, Hall effect, Seebeck coefficient, thermal diffusivity, and heat capacity measurements were also conducted on selected samples between 80 and 800K. [8]. The carrier density was calculated from the Hall coefficient, assuming a scattering factor of 1.0 in a single carrier scheme, by n or $p = 1/R_{He}$, where p and n are the densities of holes and electrons, respectively, and e is the electron charge. The Hall mobility (µ_H) was calculated from the Hall coefficient and the resistivity values by $\mu_H = R_H/\rho$. Errors were estimated to be \pm 0.5% and \pm 2% for the resistivity and Hall coefficient data, respectively. The Seebeck coefficient (a) of the samples was measured on the same samples used for electrical resistivity and Hall coefficient measurements using a high temperature light pulse technique [9]. The error of the Seebeck coefficient measurement was estimated to be less than ± 3%. The heat capacity and thermal diffusivity were measured using a flash diffusivity technique [10]. The thermal conductivity (λ) was calculated from the experimental density, heat capacity, and thermal diffusivity values. The overall error in the thermal conductivity measurements was estimated to be about $\pm 10\%$.

Results and discussion

After synthesis, MPA showed that the compositions of the Fe- and Ag-doped samples were Re_{5.56}Fe_{0.44}Te₁₅ and Re₆Te_{14.6}Ag_{0.4}, respectively. It seems therefore that, a priori, Fe atoms substitute for Re while Ag atoms substitute for Te. The room temperature properties of these samples as well as those for undoped Re₆Te₁₅ and the Re₆Te_{12.75}Se_{2.25} solid solution [6] are reported in Table I. The Re₆Te₁₅ and the Re₆Te_{12.75}Se_{2.25} samples are p-type whereas Ag- and Fe-doped samples are n-type. The room temperature properties of these samples are radically different. The hole mobility is low for p -type materials whereas reasonably high electron mobility are obtained for n-type materials. The electron mobility of the Agdoped sample is 209 cm²/Vs at a carrier concentration of 1.13 x 10¹⁷ cm⁻³. This is comparable to values obtained for n-type CoSb₃ [11] but the thermal conductivity for the Ag doped Re₆Te₁₅ doped sample is roughly 10 times lower than for CoSb₃. Considering the high electrical resistivity of the samples, the electronic component of the total thermal

conductivity is very small and, if one neglects the bipolar contribution, the thermal conductivity values shown in Table I can be considered as being the contribution from the lattice only. The lattice thermal conductivity is very low for all samples with a room temperature of 13 mW/cmK for Re₆Te₁₅. The lattice thermal conductivity value is reduced for all ternary compositions, but is substantially lower for the Agand Fe-doped samples than for the Se-doped sample. This result is somewhat surprising considering that the Se concentration is much larger than the Fe and Ag concentration in the samples. One possible explanation is that a fraction of the Fe and Ag atoms is actually being inserted inside the large voids within the crystal structure, scattering phonons efficiently. Microprobe analysis obviously does not allow to distinguish whether the Fe or Ag atoms are going into substitutional or interstitial positions and detailed x-ray analysis should be performed to determine the location of the doping atoms.

Table I. Some properties at room temperature for Re_6Te_{15} (1), $Re_6Te_{12.75}Se_{2.25}$ (2), $Re_{5.56}Fe_{0.44}Te_{15}$ (3) and $Re_6Te_{14.6}Ag_{0.4}$ (4)

	1	2	3	4
Property				
Seebeck coefficient (μV/K)	+500	+110	- 120	-10
Electrical resistivity (mΩcm)	8100	200	74	264
Hall mobility (cm ² /Vs)	3	13	6	209
Hall carrier concentration (x 10 ¹⁷ cm ⁻³)	20	22	126	1.13
Thermal conductivity (mW/cmK)	13	10	5.5	7.71

The variations of the electrical resistivity. Hall mobility, and Seebeck coefficient values are reported as a function of temperature in Figures 2, 3, and 4, respectively. With the exception of the Se-doped sample, the electrical resistivity decreases with increasing temperature suggesting that both electrons and holes participate in the electrical conduction. For the Fe-doped sample, the electrical resistivity reaches a minimum value of about 10 mQcm at 700K while the Seebeck coefficient values are nearly temperature independent and reasons for this behavior are unclear at present. The carrier mobility also decreases with increasing temperature. The power factor values (α^2/ρ) are relatively low but might be improved for samples with higher doping level. Efforts are currently in progress to prepare and characterize samples with higher electron concentrations (for example by double doping with Fe-Ni or Fe-Co). For the Ag-doped sample, the relatively high electrical resistivity and the small Seebeck coefficient values also indicate mixed electron and hole conduction. As we mentioned before, the Hall mobility values are relatively large and indicate that electrical resistivity values comparable to state-of-the-art thermoelectric materials might be achieved for extrinsic samples with a doping level in the 10¹⁹ - 10²⁰ cm⁻³ range.

Temperature dependence of the thermal conductivity is shown in Figure 5. The thermal conductivity values are compared

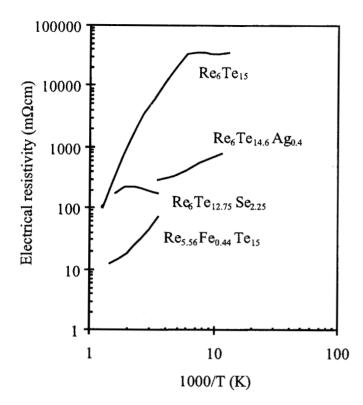


Figure 2: Electrical resistivity versus inverse temperature for some Re₆Te₁₅-based materials

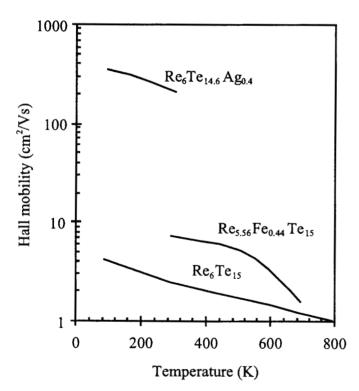


Figure 3: Hall mobility versus temperature for some Re₆Te₁₅-based materials

to those of state-of-the-art Bi₂Te₃-based alloys. At room temperature, the thermal conductivity for Re₆Te₁₅ is about 13 mW/cmK and is comparable to that for p-type Bi₂Te₃-based alloys.

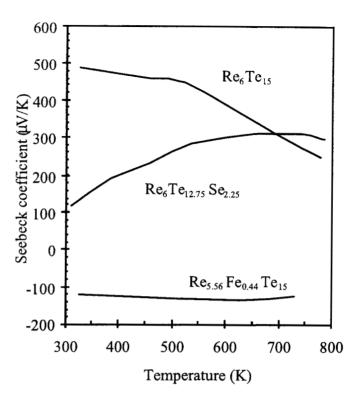


Figure 4: Seebeck coefficient versus temperature for some Re₆Te₁₅-based materials

The thermal conductivity of Re₆Te₁₅ decreases with increasing temperature following reasonably well a 1/T dependence, as expected for phonon-phonon umklapp scattering. A minimum of 7 mW/cmK is reached at 800K, significantly lower than the values obtained for state-of-the-art thermoelectric materials. Re6Se225Te1275 solid For the solution. the thermal conductivity decreases with increasing temperature approximately as T^{-1/2}. This temperature dependence is typical of phonon scattering by point defects. The values for the solid solution are lower than for the binary compound because of the mass and volume fluctuations introduced by the substitution of Se atoms for Te atoms. At room temperature the thermal conductivity is 10 mW/cmK, decreasing to a minimum of 6 mW/cmK at 600K. For the Fe-doped sample, the thermal conductivity values are nearly temperature independent with a value of 5.5 mW/cmK. This suggests that, in addition to point defects, other phonon scattering mechanisms are present. As we mentioned before, it is possible that a fraction of the Fe atoms are in fact inserted in the voids and able to rattle inside these large cavities and therefore scatter efficiently the phonons.

The minimum calculated lattice thermal conductivity for Re₆Te₁₅ is 2.3 mW/cmK at 300K [6]. At room temperature, the measured value for the Fe-doped sample is 5.5 mW/cmK, almost 3 times lower than the values obtained for Bi₂Te₃-based alloys and close to the calculated theoretical minimum. This remarkably low value suggests that figures of merit at least comparable to state-of-the-art thermoelectric materials could be achieved providing that the electrical properties can be optimized. Future studies are necessary to fully assess the potential of n-type Re₆Te₁₅-based materials as well as of filled compositions.

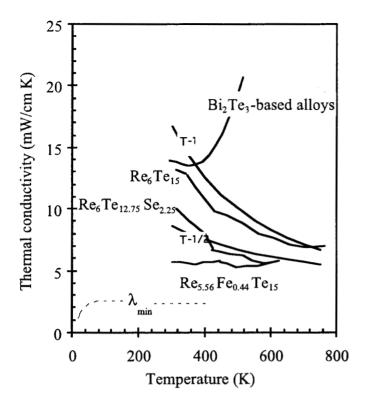


Figure 5: Thermal conductivity versus temperature for some Re₆Te₁₅-based materials. The calculated minimum lattice thermal conductivity is also reported (see reference [6] for details of calculations). Values for p-type state-of-the-art Bi₂Te₃ thermoelectric materials are also reported for comparison.

Summary

N-type Re₆Te₁₅ samples were prepared by doping with Fe and Ag. The samples are characterized by carrier mobility values substantially larger than for p-type materials. They also possess very low thermal conductivity values, significantly lower than those obtained for Bi₂Te₃-based alloys. The thermal conductivity is nearly temperature independent which suggests that a fraction of the doping atoms are inserted inside the large voids in the crystal structure, acting as efficient phonon scattering centers. These very low lattice thermal conductivity values combined with reasonably large electron mobility values demonstrate the good potential of these materials for thermoelectric applications. Further studies are however needed to determine if the electrical properties can be further optimized and the doping level controlled. In addition, attention should be paid to void filling as a way to control the thermal and electrical properties of these materials.

Acknowledgments

The work described in this paper was carried out at the Jet Propulsion Laboratory/California Institute of Technology, under contract with the National Aeronautics and Space Administration. This work is supported by the U. S. Defense Advanced Research Projects Agency, Grant No. E407.

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